

What is the Purpose of Coatings

- Coatings provide a barrier between the electrolyte (water or soil) and metal surface, providing corrosion protection
- Coatings are a primary corrosion protection method. Cathodic protection, another mitigation method, works best when paired with a bonded dielectric coating.
- Coatings must provide a defect free surface for the entire structure



Definitions

- Coating Failures Premature coating failure due to errors during surface preparation, coating application, environmental conditions, or complex problems even specialist don't know about.
- Service Conditions Premature failures due to short service life
- Coating Degradation The natural process of coating aging





Types of Coating Failures

<u>Inadequate Surface</u> <u>Preparation</u>

- Surface cleanliness
- Surface profile
- Dust/ particulate contaminants
- Lead to adhesion failures

Environmental conditions

- Amine Blush
- Condensation/ precipitation
- Cold temperature application

<u>Application</u>

- Too thin
 - Pinpoint corrosion/ rust rashing
 - Edge corrosion
 - Bridging
- Too thick
 - Solvent entrapment
 - Crazing/ cracking/ mud cracking
 - Wrinkling
 - Internal Stress
 - Runs/ Sags/ Drips

- Recoat windows
- Contaminants air and surface
 - Bubbles/ air voids
 - Fish eyes/ Cratering
 - Blistering
 - Osmotic/ salt contamination
 - Solvent
 - CO₂ off-gassing
- Materials issue
 - Crawling
 - Bleeding
 - Cold wall effect/ heat sink phenomena
 - Off ratio/ iso spit

Flaking Paint Chips due to Painting over Particulate Contaminants

- Probable Cause
 - Application of coating to loose material/ particulates
- Mitigation
 - Remove chalking/ coating materials, sand and wipe clean
- Prevention
 - Sand and wipe clean



Amine Blush

Probable Cause

- Amines have a secondary reaction with CO2 and moisture
- exasperated at low temperature and high humidity

Mitigation

 Remove loose coating, clean with detergent to remove amine blush, and sand intact coating prior to receiving a new coat



Prevention

 Use dehumidification equipment and/or heat



Flash Rust due to Environmental Conditions

Probable Cause

 Flash rust due to increase in humidity levels (ex. one pipe had cold water flowing through the pipe)

Mitigation

 Abrasive blast clean once humidity is controlled

Prevention

- Use dehumidification equipment and/or heat
- Dewater both pipes





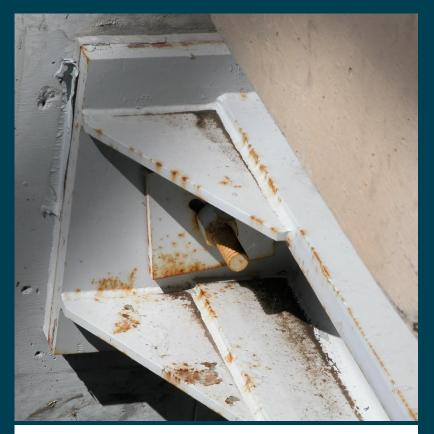
Edge Corrosion

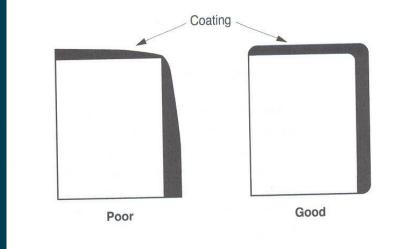
Probable Cause

 Sharp edges: coatings have a natural tendency to pull away from sharp edges

Prevention

- Grind or chamfer the edges
- Stripe coat to add more coating on edges







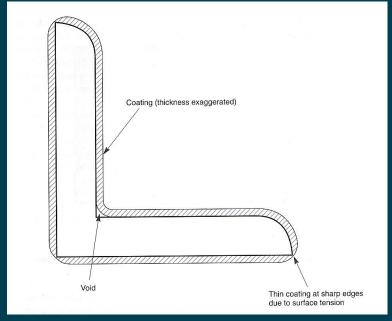
Bridging in Corners

Probable Cause

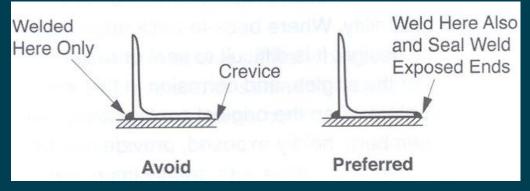
 Coating was bridged during application, crack in coating developed over a weld

Prevention

 Stripe coat using a brush to work the coating into the irregular shape









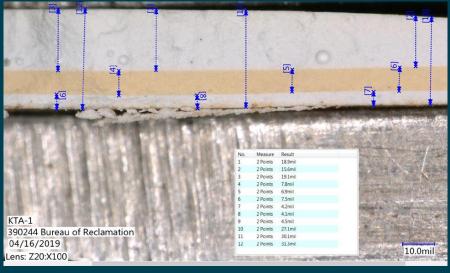
Solvent Entrapment/ Cracking/ Corrosion Rash

Probable Cause

- Excessively thick coating applied, resulting in solvent entrapment
- Increased internal stress, resulting in cracked coating

Prevention

- Good quality control
- Applicators should apply coatings within the manufacturers dry film thickness guidelines









Wrinkling, Waves, Drips, Sags, Runs

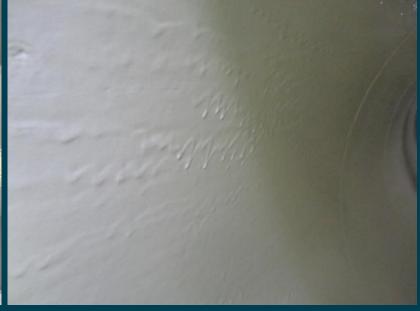
Probable Cause

 Excessively thick coating applied, resulting in runs, sags, waves, and drips

Prevention

- Coating applicator needs more practice in applying coatings, or
- Lower the volume of coating coming out of the spray gun







Heat Sink Phenomena

Probable Cause

- Observed with fast cured Polyurethane's or Polyurea's
- Coating is applied on cold pipe, but contractor is using heat and DH to control the environment
- Initial layer of paint on the steel reaction is slower than the coating in contact with the air, resulting in blistering

Prevention

 Reduce the temperature so that there is not a large temperature differential, or switch coatings for a slower reaction rate material





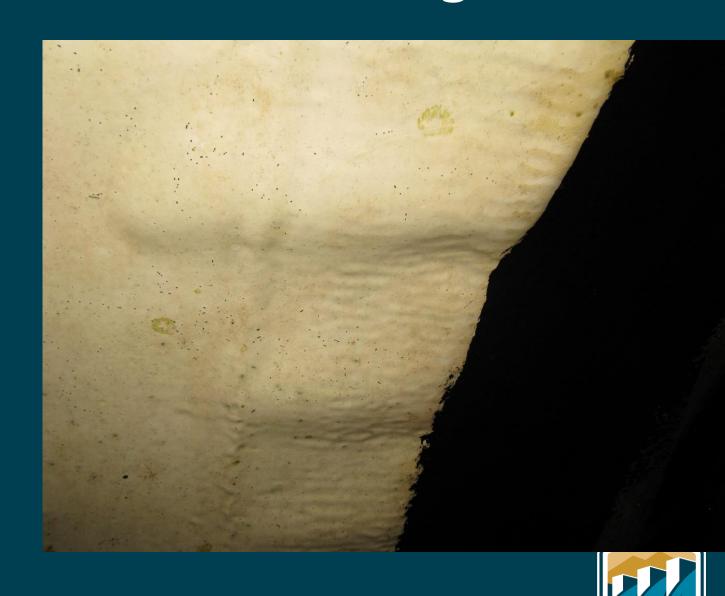
Disbondment due to Off-Ratio Mixing

Probable Cause

 Plural component coatings are applied at the gun tip, if equipment has issues, the coating ratio is not correct and will form blisters

Prevention

 Applicators must purge material to make sure materials are on ratio prior to application on structure



Service Conditions Causing Premature Failures

- Up lift forces- cause delamination
- Erosion
- Impact
- Cavitation
- UV degradationcauses chalking/ loss in gloss/ yellowing

- Slight movement/ vibrations- cause cracking
- Rivets and fastenerscan be a source of cracking
- Cyclic wetting/drying





Disbondment due to Uplift Forces







Uplift Forces (cont.)





Erosion of Linings





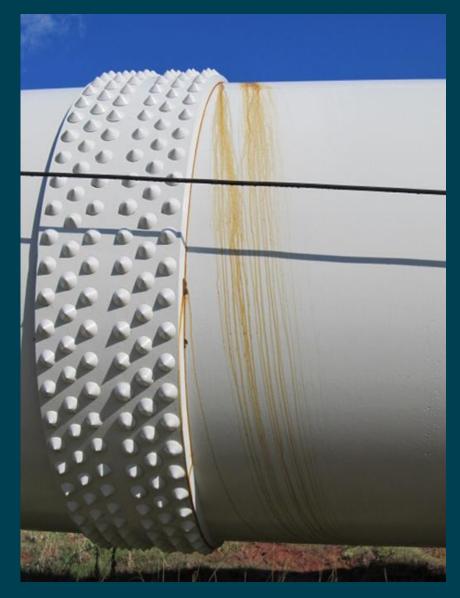


Cavitation Damage to Linings





Cracking due to Vibration







Natural Coating Degradation

- UV degradation/ chalking/ loss in gloss/ yellowing
- Blistering
- Cracking/ due to volume loss/ internal stress
- Barrier breakdown



UV Degradation







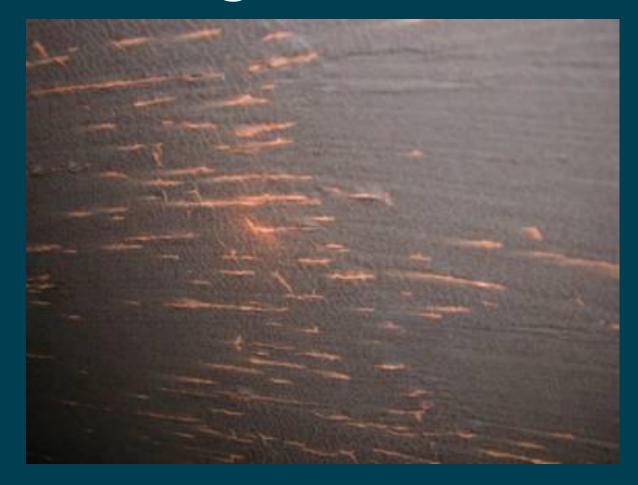
Blistering due to Degradation







Cracking of Coal Tar Enamel



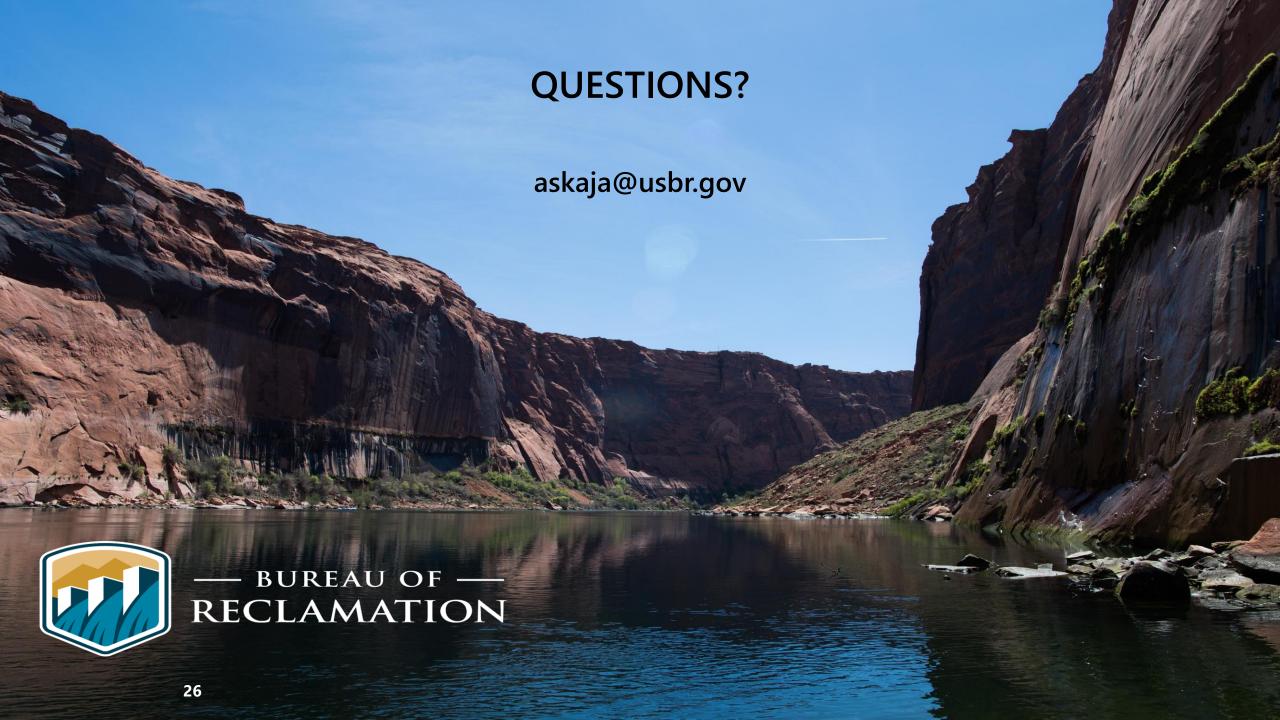




Barrier Properties Degradation







Materials and Corrosion Laboratory Staff - 8540

Cathodic Protection



Chrissy Henderson, Ph.D., P.E. chenderson@usbr.gov 303-445-2348



Matt Jermyn mjermyn@usbr.gov 303-445-2317



Daryl Little, Ph.D. dlittle@usbr.gov 303-445-2384



<u>David Tordonato, Ph.D., P.E.</u> dtordonato@usbr.gov 303-445-2394



Grace Weber
GWeber@usbr.gov
303-445-2327

Hazardous Materials



Lise Pederson, P.E. Ipederson@usbr.gov 303-445-3095



Kevin Kelly, Ph.D KKelly@usbr.gov 303-445-7944

Group Manager



Jessica Torrey, Ph.D.,P.E jtorrey@usbr.gov 303-445-2376

Protective Coatings

Brian Baumgarten bbaumgarten@usbr.gov 303-445-2399



Carter Gulsvig cgulsvig@usbr.gov 303-445-2329



Bobbi Jo Merten, Ph.D. bmerten@usbr.gov 303-445-2380



Rick Pepin, PCS rpepin@usbr.gov 303-445-2391



Stephanie Prochaska sprochaska@usbr.gov 303-445-2323



Allen Skaja, Ph.D., PCS askaja@usbr.gov 303-445-2396



